

## Mixed Ionic and Covalent Compound Naming #5

Name 'em:

- 1)  $\text{Br}_2$  \_\_\_\_\_
- 2)  $(\text{NH}_4)_2\text{SO}_4$  \_\_\_\_\_
- 3)  $\text{HNO}_2$  \_\_\_\_\_
- 4)  $\text{Pb}(\text{CO}_3)_2$  \_\_\_\_\_
- 5)  $\text{N}_2\text{O}$  \_\_\_\_\_
- 6)  $\text{ZnF}_2$  \_\_\_\_\_
- 7)  $\text{CCl}_4$  \_\_\_\_\_
- 8)  $\text{H}_2\text{S}$  \_\_\_\_\_
- 9)  $\text{Al}(\text{OH})_3$  \_\_\_\_\_
- 10)  $\text{K}_2\text{O}$  \_\_\_\_\_

Formulas, please:

- 11) sulfur trioxide \_\_\_\_\_
- 12) tin(II) sulfide \_\_\_\_\_
- 13) magnesium acetate \_\_\_\_\_
- 14) hydrophosphoric acid \_\_\_\_\_
- 15) cadmium (II) chloride \_\_\_\_\_
- 16) iron(III) phosphate \_\_\_\_\_
- 17) dinitrogen trioxide \_\_\_\_\_
- 18) chlorine \_\_\_\_\_
- 19) vanadium(V) bicarbonate \_\_\_\_\_
- 20) sulfuric acid \_\_\_\_\_